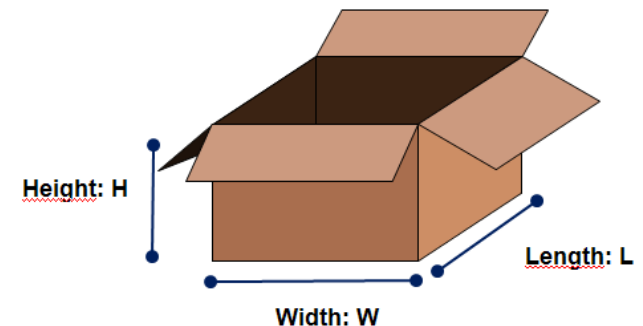


## ***DO NOW:***

**Date: November 6, 2017**

***TEKS: 6.6(B) Calculate density to identify an unknown substance***

1. Get out your **Do Now sheet** and Mystery Powders Lab paper
2. Write this week's TEKS onto your **Do Now** sheet.
3. Write Monday's question on your **Do Now sheet** and answer it. **Q?: How do you calculate the volume of a rectangular box?**



## ***DO NOW:***

**Date: November 7, 2017**

***TEKS: 6.6(B) Calculate density to identify an unknown substance***

1. Get out your **Do Now sheet**
2. Write Tuesday's question on your **Do Now sheet** and answer it! **Q?: Which is more dense: liquid water or an ice cube? Explain your reasoning.**

***Be ready to be cold called!***



## ***DO NOW:***

Date: November 8 – 9, 2017

***TEKS: 6.6(B) Calculate density to identify an unknown substance***

1. Get out your **Do Now sheet**
2. Write Block Day's question on your **Do Now sheet** and answer it! **Q?: How do you find the density of an irregular solid?**



## ***DO NOW:***

Date: November 10, 2017

***TEKS: 6.6(B) Calculate density to identify an unknown substance***

1. Get out your **Do Now sheet**
2. Write Block Day's question on your **Do Now sheet** and answer it! **Q?: One gram (1g) of water has a volume of one cubic centimeter (1 cm<sup>3</sup>). What is the density of water?**

$$\text{Density} = \text{mass} \div \text{volume}$$

